

Navgan Shikshan Sanstha Rajuri's(N)

MRS.KESHARBAI SONAJIRAO KSHIRSAGAR ALIAS KAKU

Arts, Science And Commerce College, Beed.

NAAC reaccredited with 'A' grade (3.18 CGPA as per new RAF) & ISO 9001-2015 certified





7.2.1 Some more Best Practices of the institution:

- 1. Wallpapers- Publications and Exhibitions
- 2. Student mentoring system
- 3. Remedial classes for slow learners
- 4. Counseling and Guidance Committees for Admission Process
- 5. College magazine 'DURWANKUR'
- 6. 'Science exhibition
- 7. 'One Night for Senior Citizens' program
- 8. 'School-College Students interactions'
- 9. National Chemistry Quiz Competitions

Best Practice-1

Title of the practice: Wallpapers- Publications and Exhibitions

Objective of the practice:

The main objective of this practice is to involve U.G. & P.G. students and motivate them to current scenario of innovations and discoveries around the world. The purpose is also to involve the students for sharing original research, contributing to academic knowledge and building their bright career. It can also assist to improve their critical thinking, writing skills and provide opportunities for networking and collaborations with other expertise.

The context:

During the formal graduation program, the students do not get enough exposure to enhance their critical thinking, logical reasoning, research and innovative skills so as to build upon strongly for their future bright trajectory. Moreover, most of the students of our college belong to the rural and backward area so the students needed to be propagated in the right direction.

4. The Practice:

The college initiated this practice 25 years ago since then; students are continuously motivated to participate in wallpaper preparation, publication and its exhibition. On every Independence Day and republic day, wallpapers on the current issues and innovations around the world are prepared and published. The students of around 22 departments of the college participate in this activity regularly. All the students are notified in advance well before 15 days to prepare wallpapers. Interested students register their names to the concerned department and then the students are advised, guided, monitored and inspired to prepare wallpapers on various burning issues of the society, literature, culture, science and environment. Each department suggests a particular theme based on which the students prepare their wallpapers. Students are provided with appropriate study material, required infrastructure, ICT facilities and some financial assistance. Students make a group of 2 to 5 students based on the theme they are interested in. Each group of students is monitored and guided by a teacher. After taking final review of wallpapers by the

corresponding teachers, these wallpapers are then published at the auspicious hands of a guest of honour for the flag hosting on the Independence Day and the Republic Day in the college campus. After publication, these wallpapers are then demonstrated in the exhibition. Demonstrating students explain the content of wallpaper to the many other students and interact with them on the theme of the wallpaper.

Evidence of Success:

- This practice helped the students to reduce their fear of interaction.
- > It enhanced the communication skills of the students.
- ➤ It also inculcated skills like presentation skill, discussion skill, listening skill, critical thinking, and writing skills among students.
- Students are able to show competence in identifying relevant information, defining and explaining themes under discussion.
- ➤ Students are able to judge when to speak and how much to say, speak clearly and audibly.
- ➤ It increased students' excitement and interest and confidence towards participative projects.
- ➤ It helped student to be aware of recent innovations and discoveries.

Problems encountered and recourses required:

The problems encountered:

- Since many students visit to the exhibition, it is quite difficult for presenter students to give proper justice to the theme.
- > Sometimes presenters get tired due to consistently interacting with students.
- ➤ Poor communication skill can impact exhibition.

Resources Required:-

- > Appropriate budget allocation
- > Study material on current discoveries
- ➤ Enough ICT facilities, computers, internet facility.

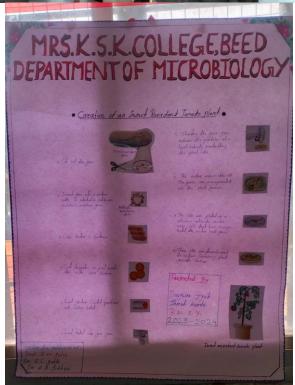
Photo of Wallpaper Publications:

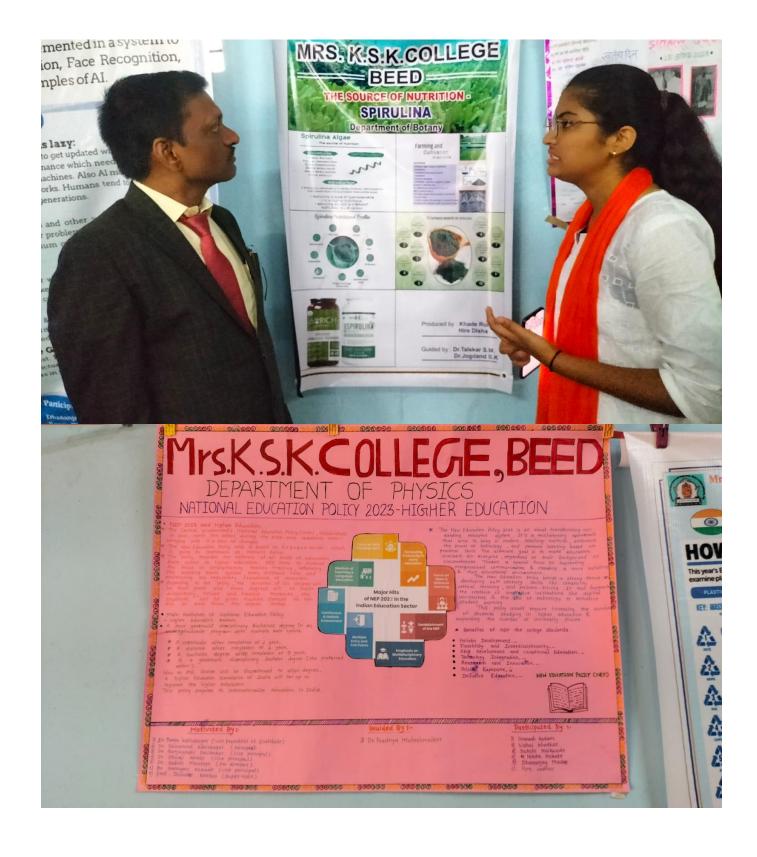






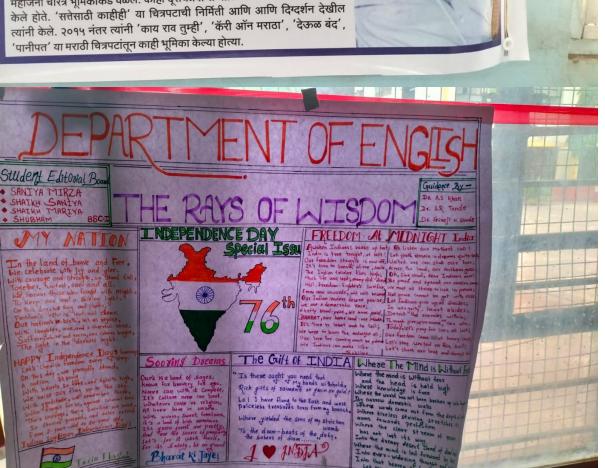








रविंद्र महाजनी यांना मधुस्दन कालेलकर यांच्या 'जाणता अजाणता' या नाटकातून पहिली संधी मिळाली आणि त्यांनी त्यात साकारलेली मुख्य भूमिका गाजली. नंतर कालेलकर यांनी खास त्यांच्यासाठी 'तो राजहंस एक' हे नाटक लिहून काढले. व्ही. शांताराम यांनी या नाटकाचा एक प्रयोग पाहिला आणि महाजनींना 'झुंज' या चित्रपटातील मुख्य भूमिका दिली. १९७४ मधील हा चित्रपट तेव्हा चांगला गाजला. या चित्रपटाने त्यांचा नऊ वर्षांचा संघर्ष संपवला. 'झुंज' चित्रपटाने रौप्यमहोत्सवी यश मिळवले आणि महाजनीं मराठी चित्रपट सृष्टीतील मोठे कलाकार झाले. त्यानंतर 'आराम हराम आहे', 'लक्ष्मी', 'लक्ष्मीची पावलं', 'देवता', 'गोंधळात गोंधळ', 'मुंबईचा फौजदार' या व इतर चित्रपटात महाजनींनी काम केले. त्या काळात तमाशापटांची आणि प्रामीण पार्श्वभूमीच्या चित्रपटांची चलती असताना खास महाजनींसाठी शहरी रूपाला शोभून दिसतील, अशा कथा लिहिल्या गेल्या. १९७५ ते १९९० या काळात महाजनी मराठी चित्रपटातील मोठे कलाकार म्हणून प्रसिद्ध होते. महाजनींनी 'बेलभंडार', 'अपराध मीच केला' या मराठी नाटकांचे अनेक प्रयोग देखील केले. इ.स. १९९० नंतर महाजनी चिरेत्र भूमिकांकडे वळले. काही दूरचित्रवाणी मालिकांमध्ये देखील त्यांनी काम केले होते. 'सत्तेसाठी काहीही' या चित्रपटाची निर्मिती आणि आणि दिग्दर्शन देखील केले होते. 'सत्तेसाठी काहीही' या चित्रपटाची निर्मिती आणि आणि दिग्दर्शन देखील केले होते. 'सत्तेसाठी काहीही' या चित्रपटाची निर्मिती आणि आणि दिग्दर्शन देखील

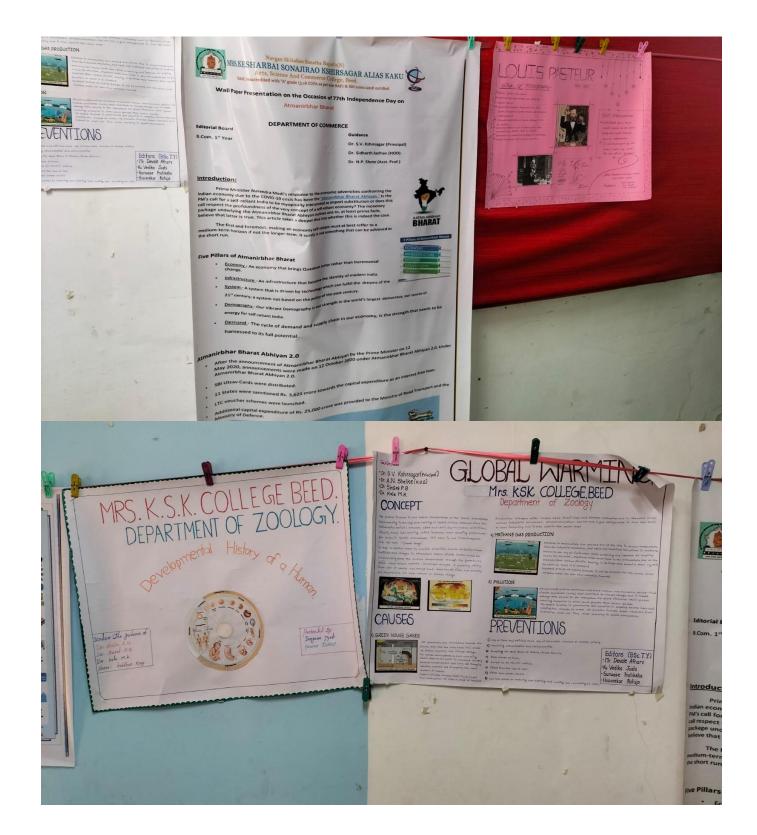


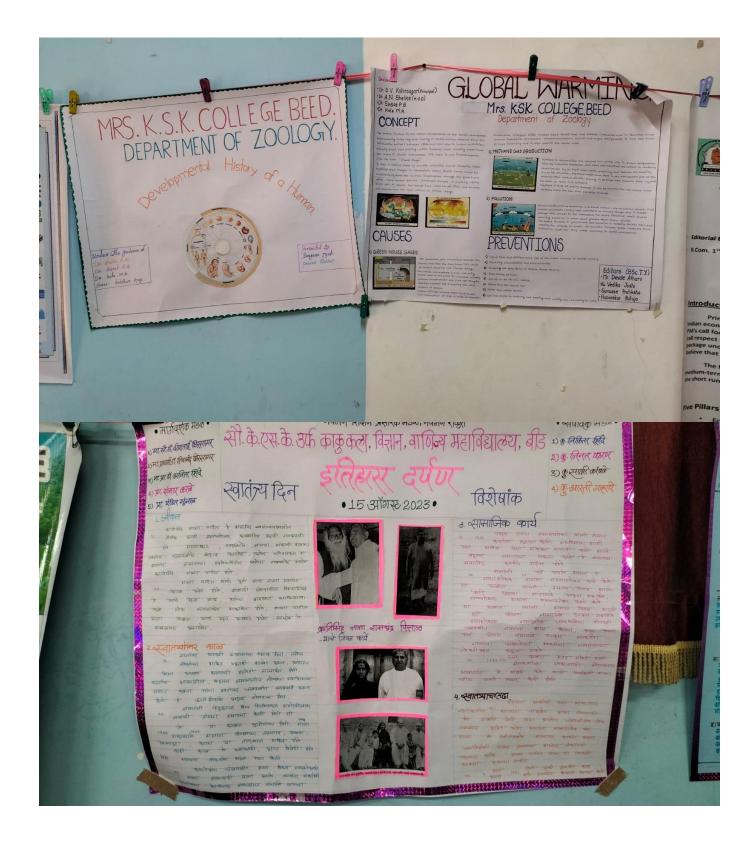


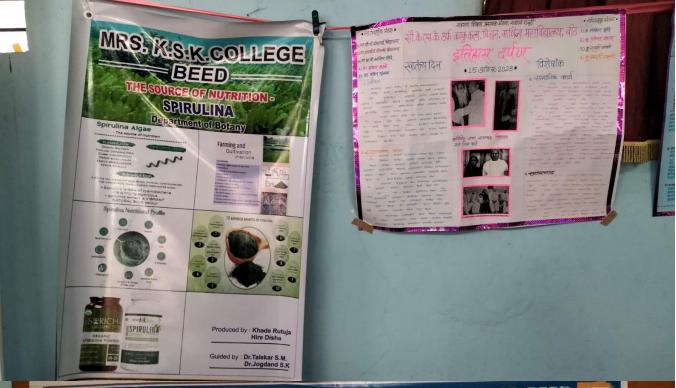
States, chance only favors the mind which is prepared:

NAMES - Kushikesh Mahendea +
- BSC-MICRO - I Jene .

Guidance:







MRS. K.S.K COLLEGE OF ARTS, SCIENCE & COMMERCE, BEED. - Department Of Computer Science. -

Artificial Intelligence

Al is a broad field which refers to the use of technologies to build machines & computers that have ability to mimic cognitive functions associated with such as being able to see, understand & respond to spoken or written language, analyze data, makes recommendation and morel. Although AI is often thought of as a system in itself, it's a set of technologies implemented in a system to enable it to reason, learn & act to solve a complex problems. ChatGPT, Self Driving Cars, Speech Recognition, Face Recognition, Image Generator, Google Maps, Industrial Robots, Text Editor & Virtual Assist these are some of trending examples of AI.

- Pros Of AI-

Reduction Of Human Error & Medical Applications:
The phrase "Huma Bror was born because it make mistakes time to time. However computers domaines these mistakes if they are programmed properly. At has also made significant contribution to the field of medicine, with applications ranging from disgnossis & treatment to drug discovery & clinical traits. At Powered tools can helps doctors analyze patient data, identify potential health risks andeelop personalized treatment plans.

health risks and develop personalized treatment plans. Take risk; instead of Humans:

This is one of the biggest advantage of Al We can overcome many risky limitations of humans by developing and Robots which in turn can do the risky things for us. Let it be going on mars, defuse a bornh, explore the deepest part of oceans, mining for coal & oil, it can be used effectively in any kind of natural or man made disaster.

of naturator man-mans.

Available 24x7:

An average human will work for 4-6 hours a day excluding the breaks. Humans are built in such a way to get some time out for refreshing themselves 8 get ready for a new day of work. But using AI we can make machines works 2607 without any breaks 8 they don't even get bored, unlike Humans.

Taster Decisions:

Using Al alongside other technologies we can make machines take decisions faster than a human & carry out actions quicker. While taking a decision human will analyze many factors both emotionally & practically but Al powered machine works on what it programmed & delivered the

Al image generator we can generate the image as we want. We can make image of that nary person using Al by describing their how the y looks like, their attire, their skin toon, a ste many more

- Cons Of AI -

High Costs of Creation & Making Humans lazy:

As Al updating every day the hardware & software need to get updated with time to meet the latest requirements. Machines need repairing & maintenance which need plenty of costs. It's creation requires huge costs as they are very complex machines. Also Al making humans lazy with it's applications automating the majority of the works. Humans tend to get addicted to these inventions which can cause a problems to future generations.

Oneenproyments.

As AI replacing the majority of the repetitive tasks and other works with robots, human interfacing is becoming less which will cause a major problem in the employment standards. Every organization is looking to replace the minimum qualified individuals with AI robots which can do similar works with more efficiency.

No Emotions:
There is no doubt that machines are much better when it comes to working efficiently buy they can't replace the human connection that makes the team. Machines can't develop a bond with humans which is an essential attribute when comes to team Management.

No Human Replication:

Humans have created machines to save time & efforts from doing non-essential repetitive
tasks. Al-powered machines works on algorithms, mathematical computing & cognitive
technologies. They can become highly advanced but can't act or think like a humans.

Encourage Inappropriate Image Generation:

We all know Al image generator is the best But it can also create or design inappropriate image of a person, it can misuse of our pictures and convert that image into inappropriate subject that we shouldn't consider to discuss in publicity.

- Motivated By D. B. Kshirsagar

Participate Students

प्रेरणास्यान : इं. दीपा सीरसागर

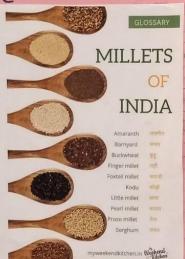
प्रान्पार्थः डॉ.शिवानंद क्षीरसागर उपप्रान्यार्थः उ. संजय पाटील उपप्राचार्यः डॉ. शिवाजी शिंदे फर्मावे उपप्रान्यार्थ : डॉ. नारायण काकडे पर्यवेसकः प्रा. जातिंदर कीळेकर वापर भरड धान्याऱ्या

हरित फ्रांतीमुळे देशान्या अत्नधान्य उत्पादनात वरीन्य वाढ झाली:माल त्यामुळे शेलकरी पारंपारिक पिकांपेक्षा नगदी पिकांकडे अधिक वळू त्यागत्या. गेत्या काही वर्षीपासून जागतीक हवासान पालटाचा फटका शैतकऱ्यांना मोख्या प्रमानात बसत आहे. 'आध्निक जीवनशैभी' च्या नावायात्मी अनेकांच्या आहाराच्या सवयीत पालट झालेले आहेत. संयुक्त राष्ट्रांनी वर्ष 2023 हे 'आंतरराष्ट्रीय मरड थान्य वर्ष म्हणून बोषित केले आहे. झरड धान्याल ज्वारी, बाजरी , राजिशा, कोद्रो, कुटकी , शावा, राळा, वर्ख (भगर) नान्यणी या पिकांन्या समावेश होतो.

विववणे आवश्यक आहे

भेर्डधान्यः काळाची आवश्यकता

हशास्त्र विभाग



डॉ. माया आंदार मंजु जाधव विद्यार्थी संपादक श्वेता थ्वरात (11वी कला) शिंद अंजली, शेख फिरोजा

संपादक मंडळ

असून ती 'पीरिक तृणधान्य म्हणून ओळखत्मी जातात आणि पचायला हत्यकी असातात, त्यामुळे भूमीतील नकाचे प्रमान वादून भूमीन्या पोत सुधारतो ही धान्ये पठ्यांन्ये आवडते आद्य असून जैवविविद्यता वाढते.तसेच थांपासून विविध पदार्थ सिद्द ५रून विकल्याने अन्यत गट स्वयंपूर्ण होजा शकुतात

भरव धान्य पिकांन्ये शुणधर्म 🚓

बाजरीमध्ये फॉस्फरस उच्च असाणात असून, ते वेशींसधील उर्जा आणि अन्य अतिज पदार्थ साठपण्यास साराय्य करतात. त्योहान्येही प्रमाण अधिक असल्यामुळे शरिरातील हिमोञ्लोबीन याउने. नाचणीमच्चे नेसर्जिङ डॅल्बिजन सर्वाधिक असल्वासुळे अतिदिनन्या सेवनासुळे हारांचे आरोज्य

जागितिक भर्ड धान्य उत्पन्नाच्या ४९ टक्के भरड धान्य पिकांची वैशिष्ट्ये स्रुणजे ही अल्प चांगले राहते. राळे हे पाचक असून त्यात लोह आणिअलि उत्पन्न भारतात होते आहारातील सहत्त्व जाणून पाक्साच्या परिस्थितीत शुष्ठ आणि अर्ध-शुष्ठ भागान पदार्घ सरपूर आहेत वर्डमध्ये उन्ज लोह धानू आहे. कोडो घेउन्त त्यांच्या वापर दैत्तंदिन आहारासध्ये होणे वाढमान थांपासून ॲलर्जीच्या धोका नाही. या पिकांसध्ये केंद्रो मन्नासंस्था मनवूत करण्यासाठी उत्हब्द आहे. असा अत्यंत आवश्यक आहे. थासाठी था धान्याचे रासायनिक अने आणि किटकणाशके यांचा वापर अन्य प्रकारे झारतीय भरड धान्याचे सहत्त्व जाणून शेतकचांनी लाक्न जनसामान्यांच्या सलावर विविद्य स्तरांत्र प्रमाणात होतो. पर्यायाने उत्पादन व्यय अल्प होजन नगदी विकांच्या सात्रे न लागमा ही पारंपारिक विके वेजन उत्पादनात वाढ होते. या पिकांसच्ये उच्च पीव्हिक सुल्ये देश सुजलाम् सुफलाम् करावा







Best Practice-2

Title of the practice: Student mentoring system:

Goal:-

For the overall development of students and improve the interaction between teachers and students.

The Context:

The current class teaching scenario wherein only the class teacher alone is in contact with the student of respective class to some more extent. However rest of the faculties have not enough interaction with the students. To overcome this barrier, the college has introduced the student mentoring system to fulfil the following tasks.

- To take interest in developing students career and well being.
- To have interpersonal and professional relationship with mentees.

Some faculty limit the responsibilities of mentoring to simply discharging their role as advisor however, assigned advisors are more effective who play important role in development of future of students. The role of advisor is limited to guiding academic progress but the role of mentor is focused on advancing students career through interpersonal relation that facilitate sharing guidance & experience.

The practice:-

Each mentor is allotted with 40-50 students. The students allotted to each mentor on the basis of subject and classes handled by the mentor. The care was taken that each student will have the same mentor for the next three years of his graduation. The proforma has been designed for the personal details of the students including his family details. Each mentor is asked to prepare a details of the activities to be undertaken per semester and to distribute among the mentees allotted to him for this, mentor conducts the initial meetings so as to get information what kind of mentoring they need by asking few questions. In these meetings, mentor points out individual strength and weakness of mentees so as to plan the activities and guidance. Each mentor assess his mentees through various meetings and report his progress to guardians of mentees through meeting with them at the end of every month. Mentor also note down some constructive suggestion by parents.

Evidence of Success:-

Effective mentoring good for mentors, mentees and good for the discipline. One of the successes to implement mentoring system is the dropout rate of the college minimized. The student's attendance per class has increased remarkably.

For mentees

Student-Teacher strong bond is developed. It has helped to improve communication skills, social skills. Students felt connected and shared their difficulties.

For College

Mentoring system has been highly appreciated by the mentee's parents. The discipline has been improved in the college campus. Involvement of knowledge source in the actual education process. It has developed of good support service. It is step ahead towards student-teacher communication. It served young students in a more responsible way in College campus College has indirectly set a role model as a mentor in front of students. Mentor is introduced to the important issues of students. Mentor as a caring adult has to improve communication skills. Mentors developed insight for student's problems and stress. Mentor experience the problems of mentees and get closer to students from different socio economic status. Mentors felt, it is an opportunity to cooperate adolescents to nurture career

Resources:

Willingness of faculty members to adapt 40-50 students as their mentees for their overall development and inculcate value education among them so as to make them employable.

.... Mentee Information Form

Academic Year 2019 - 2020

Name of Mentee :	
Class :	Division :-
Roll No. :	
Optional Subject :	
	Contact Details
Permanent address :	*
Current address :	
-	
Guardian Name:	Mob.No. :
Mail address :	
Adhar No.:	PRN No:
Signature of Mentee:	
Name of Mentor :	Signature :

Mentee Report Card

Academic Year 2019 - 2020

Date / Time	Problem / Issues Solved	Sign. of Mentee	Sign. of Mentor
			,

Half Yearly Report

Academic Year 2019 – 2020

Co-Curriculum Activities

Sr.No.	Activity	No.
1	Sports competition	
2	Music competition	
3	Debate competition	
4	Essay Writing competition	
5	Poem recital competition	
6	Quiz competition	
7	Wallpaper presentation	
8	Science Exhibition	
9	Study tour visit	
	Other competition	
10	1-	
10	2-	
1	3-	
	Total number of Winners	

Problem / Issues Solved

Sr.No.	Activity	No.
1	Academic problems	
2	College campus problem	
3	Co-curricular participation problem	
4	Traveling problems	
5	Family problem	-
6	Personal Stress	

Name of Mentor:-----Signature:-----

Annual Report

Academic Year 2019 - 2020

Co-Curriculum Activities

Sr.No.	Activity	No.
1	Sports competition	
. 2	Music competition	
3	Debate competition	
4	Essay Writing competition	
5 '	Poem recital competition	
.6	Quiz competition	
7	Wallpaper presentation	
8	Science Exhibition	
9	Study tour visit	
	Other competition	
10		
10	2 -	The second second
	3 -	
	Total number of Winners	

Examination

Sr.No	Particulars	No.
1	Total number allotted mentee	
2	Number of mentee attended examination	1 1
3	Number mentee passed examination	
4	ATKT	
	Number of mentee not attended examination	

Problem / Issues Solved

Sr.No.	Activity	No.
1	Academic problems	
2	College campus problem	
3	Co-curricular participation problem	
.4	Traveling problems	
5	Family problem	1 1 1 1 1 1
6	Personal Stress	

Name of Mentor :- ------Signature :- -----

Communication Report Card

Academic Year 2019 – 2020

Faculty:Class:			
Sr. No.	Name of Mentee	Mobile No	Remarks
1)	S. M. D. Kore Stand St. 18		the water
2)			A CALL STORY
. 3)			A CALL OF THE REAL PROPERTY.
4)			
5)			
(6)			
7)	Talk to A control and		
8)		March 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	
(9)			
10)		A A Carlotte Comment	
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12)			
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19)			
(20)	May to a second		
21)			and the second second
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23)			The state of the s
24)		A LANDER DE LA CONTRACTOR DEL CONTRACTOR DE LA CONTRACTOR DE LA CONTRACTOR DE LA CONTRACTOR	476.1

Best Practice-3

1. **Title of the practice:** Remedial classes for slow learners

2. Objectives of the practice:

- To encourage the academically weaker students to identify and work on their weakness and help them to improve on their learning skills.
- To enhance the level of understanding of basic concepts in various subjects.
- To provide a stronger base for further academic progress.
- To strengthen their knowledge skills and attitude in such subjects.
- To improve academic skills of the students in various subjects.

3. The Context:

One of the objectives of our college is to provide good quality education to students from rural areas and to develop their interest in education. Beed district in Maharashtra is considered to be underdeveloped and backward in terms of development. Almost half of the students admitted to our college are from underprivileged society of rural areas. Due to the lack of good educational institutions in the rural areas, the students are found to be out of the mainstream of education. Also, due to the poor financial situation of the house, they have to contribute to the work of their parents every day. Therefore, these students could not take proper education to strengthen their knowledge skills and academic skills. Hence the college has decided to conduct remedial classes in all subjects for these students so as to bring them in a mainstream of the education.

4. The Practice:

At the beginning of every academic year, a test is conducted to identify the academically slow learners. Notice regarding remedial class is circulated among the students who are aspiring and willing to attend the remedial classes. Slow learners are also identified during the regular classes. Once the slow learners are identified, a curriculum based on some basic concepts is designed by the head and professors of the corresponding departments. Systematic schedule is prepared for the delivery of content of remedial coaching classes with the consent and permission of the principal of the college. Remedial classes are organisedduring the two days of the week so as to meet the needs of the slow learners.

Scheduled content of the remedial class is delivered during two months. One of the professors in the department is given the responsibility of identifying and resolving the problems of slow learners. During the course of remedial class, tests are conducted regularly to monitor the progress of slow learners and at the end of the remedial class course, an examination is conducted so as to observe the progress of the slow learners.

5. Evidences of Success:

Due to these remedial classes, the progress of the students can be monitored and the teachers get to know the weakness and the strengths of the students so that the teachers can plan properly for their overall development. These classes solidify the basic concepts of the students and they understand the main core concepts of each subject as well. This eventually resulted in progress in their final year results. Mainly, due to these classes, confidence was shown by these students to confront any exam.

6. Problems encountered and Resources Required:

In-charge professors were allotted an extra workload to conduct remedial classes but this turned out to be an excellent piece of devotion towards these students. To conduct these classes, the overall time table of the college had to be reshuffled. Time constraint is always there to conduct these classes as it requires quite more time so as to address the different problems of the students. More ICT based content and ICT tools are required to enhance the interests of the students in the concerned subjects.

Best Practice-2:- Remedial Classes for slow learners; Photos

Remedial Coaching Course: Syllabus

B.Sc. II Year Chemistry(2020-2021) Total: 24 Hours, Three hour per Week

Chemical compounds & their families Unit - I 06 Hours

- Types of chemical compounds,
- Functional groups of organic compounds
- Different theories of Acids and Bases,

Basic concepts in Organic Chemistry: Unit - II 06 Hours

- Chemical Bonds & it's types,
- Intermolecular forces & it's types,
- Mechanism of Chemical reactions,
- Hydrocarbons and types
- Heterocyclic compounds

Unit - III Spectroscopy; 06 Hours

- Brief account on spectroscopy,
- Types of spectroscopy,
- Elucidation of spectral data,

Unit - IV Laboratory skills in Performing Experiments:

06 Hours

- Operating Instruments,
- Handling Apparatus, glass wares etc.
- Preparation of Standards Solutions of different Normality, Molarity.

Department of Chemistry Mrs. K.S.K. College, Beed.

Remedial Coaching Course: Syllabus

B.Sc. I Year Chemistry (2021-2022) Total: 24 Hours, Three hour per Week

Unit –I <u>Mathematical Concepts:</u> 06 Hours

- Basic mathematics used in Physical Chemistry
- Differentiation, Integration, etc.
- Logarithmic Rules;
- Internal conversions; numbers in points to powers.
- Practice of drawing various types of graphs in Physical Chemistry,

Unit - II Physical Properties and it's Units: 06 Hours

- Measurements of Physical Properties,
- Units of Physical Properties,
- Unit Conversions; SI units to CGS units,

Unit – III <u>Laboratory skills in Performing Experiments:</u> 06 Hours

- Operating Instruments,
- Handling Apparatus, glass wares etc.
- Preparation of Standards Solutions of different Normality, Molarity.

Unit - IV Atomic Structure: 06 Hours

- History of atom, Atomic orbital's,
- Quantum numbers, periodic properties
- Electronic configurations of the elements,

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Department of Chemistry

Time - Table for the Remedial Course; 2021-2022

B.Sc. I, II Year.

Day / Time	Class	Name of the Teacher	Hall No.
Monday 7.30 – 8.20 am	B. Sc. II Yr.	Dr.P.R.Khakre	50
Tuesday 7.30 – 8.20 am	B. Sc. II Yr.	Dr.P.R.Khakre	50
Wednesday 7.30 – 8.20 am	B. Sc. II Yr.	Dr.P.R.Khakre	50
Thursday 7.30 – 8.20 am	B. Sc. I Yr.	Dr.S.B.Maulage	50
Friday 7.30 – 8.20 am	B. Sc. I Yr.	Dr.S.B.Maulage	50
Saturday 7.30 – 8.20 am	B. Sc. I Yr.	Dr.S.B.Maulage	50

HOD - Chemistry
Department of Chemistry

Mrs. K.S.K. College Reed

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Department of Chemistry

Remedial Coaching Course

B.Sc. II Year Chemistry (2021-2022)

Student list

Sr.no	Name of student
1.	Akhade Nikhil Vishnu
2.	Dake Trupti Indrajit
3.	Gaikwad Kartik Sundar
4.	Ghule Omkar Govindrao
5.	Kale Akshay Sadashiv
6.	Late Ajit Parmeswar
7.	Pathak Gajanan Girishkumar
8.	Phad Santosh Manik
9.	Rajpure Archana Bandu
10.	Sable Umesh Laxman
11.	Shinde Meena Namdev
12.	Shinde Vaibhavi Ashruba
13.	Sirsat Rohit Vilas
14.	Sonwane Mayur Bhaskar
15.	Survase Aishwarya Babasaheb
16.	Takale Omkar Hanuman
17.	Tandle Prachi Rajabhau
18	Thokal Sakshi Santosh
19.	Thokal Vishal Babasaheb
20.	Thombre Rushikesh Murlidhar
21.	Vaidya Kalyani Prerak
22.	Varad Bhagyashree Santosh
23.	Waghmare Pormina Navnath
24.	Wahed Shaikh Malik Wahed Shaikh
25.	Wakhare Vaishnavi Vitthal





24.	Salve Saurabh Rajendra	
25.	Takik vaishnavi Sham	
26.	Yadav Jay Vijay	

Head
Department of Chemistry
Mrs. K.S.K. College, Beed

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Department of Chemistry

Remedial Coaching Course

B.Sc. I Year Chemistry (2021-2022)

Student list

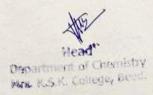
Sr.no	Name of student	
1.	Aghav Aarti Arjun	
2.	Akhade Sagar Satish	
3.	Barde Karan Kailas	-
4.	Chate Shubham Bappasaheb	
5.	Dhas Tushar Khanderao	
6.	Dhase Pavan Ankush	
7.	Gadale Ankush Janak	
8.	Garud Aniket Ganesh	
9.	Ghuge Rutuja Balu	
10.	Jadhav Pratik Balasaheb	
11.	Joshi Kanchal Dilip	
12.	Kale Nikhil Datta	
13.	Kamble Rahul Sunil	
14.	Kate Ganesh Bharat	
15.	Kekan Gorakh Dhondiba	
16.	Korde Nishant Balu	
17.	Kute Nitin Babasaheb	
18	Magar Anand Sanjay	
19.	Mane Pradnya Sham	
20.	Munde Sunanda Ramesh	
21.	Nagargoje Pratiksha janardhan	
22.	Pawar Avinash Sanjay	
23.	Rasal Pooja Vasant	

Department of Chemistry 2021-22

Remedial Class Test - B.Sc.II year

Marks obtained =

Dat	te:	Out of	= 20	
Mu	ltiple choice questions with one correct answer.			
1.	The general formula for alkanes is			
	a) $C_n H_{2n}$ b) $C_n H_{2n+2}$ c) $C_n H_{2n-2}$ d) None of these			
2.	Alkanes mainly show			
	a) Ionic elimination reactions b) Ionic formation reaction			
	c) Free radical elimination reactions d) Heat/Photochemical reaction	S		
3.	Which one of the following bonds is strongest?	75		
	a) - C - C - b) - C = C - c) - C \equiv C - d) None of these			
4.	Which one of the following is not found in free state?			
7-				
5.	a) Au b) Na c) Cu d) Ag Alkali metals are generally prepared by			
100				
6.	c) Electrolytic reduction of their molten halides d) Reduction of metal oxide			
U.	The decreasing order of reactivity of hydrogen atoms attached to different carbo	n atoms in aikai	ies is	
	a) Tertiary > Primary > Secondary b) Tertiary > Secondary > Primary			
7.	c) Primary > Secondary > Tertiary d) Secondary > Primary > Tertiary Iodination of alkanes is best carried out in the presence of			
1.				
0				
8.	Alkali metal have how many electron in their outer most shell			
0	a) 0 b) 1 c) 2 d) 1 or 2	er vereze		
9.	Which of the following is the most abundant alkali metal (in combined state) in	nature?		
10	a) Li b) Na c) Cs d) k			
10.	In the series, ethane, ethylene and acetylene, the C-H bond energy is			
		Equal in all		
11.	Electrochemical process is employed to extract			
	a) Mercury b) Cadmium c) sodium d) silver	epara and		
12.	The maximum ease of abstraction of a hydrogen atom by a chlorine atom is show			
	a) $(CH_3)_4C$ b) $(CH_3)_3CH$ c) $C_6H_5CH_3$ d) $CH_2 = CH_3$	HCH ₃		
13.	The reaction RX + 2 Na + RX R-R + 2NaX is called			
	a) Fittig reaction b) Sabetier and Senderson's reaction c) Wurtz's 1	reaction d)	Wurtz	fittig
	reaction			- 3
14.	A mixture of C2H6, C2H4 and C2H2 is passed through ammonical AgNO3 so	lution. The gas	ses which re	emain
	unreacted are a) C2H6 and C,H, b) C,H, and C,H, c) C,H, and	C,H, d)	None of thes	e
15.	Element of 1st group give colour in Bunsen burner due to	* * *		
10.	a) Low ionisation potential b) Low melting point c) High reactivity d) One ele	ectron in their o	utermost she	ell
16	The ashes of plants contain alkali metal, 90% of which is	cuon in their c	dicimost sin	211
10.	a) Li b) K c) Cs d) Ca			
17	Formation of alkanes by the action of Zn on alkyl iodide is called			
17.	a) Frankland reaction b) Cannizzaro reaction c) Kolbe's reaction	d) Wurtz re	action	
18	Which of the following electronic configuration represents an alkali metal?	d) waitz it	action	
.0.	a) ns ¹ b) ns ² c) ns ² np ¹ d) ns ¹ np ²			
10	Alkali metal are			
17.	a) Reducing agents b) Oxidising agents c) Both reducing and oxidising a	agents d)	Compl	avina
	agents	gomo u)	Compi	camg
20	The minimum first ionisation energy is shown by which of the following electron	nic configuration		
LU.	a) 1s ² 2s ¹ b) 1s ² 2s ² 2p ⁶ 3s ² 3p ² c) 1s ² 2s ² 2p ⁶ 3s ¹ d) 1s ² 2s ² 2p ⁶			
	a) 15 251 0) 15 25 24 35 34 0) 15 25 24 35 d) 15 25 24			



Name :

Remedial Coaching Classes (2021-22)

Introduction:

The department of chemistry has been suggested by Hon. Principal to conducts the remedial classes for the slow learners in the subject of chemistry. Therefore the chemistry department has decided to run the remedial classes for weak students/ slow learners in the subject of chemistry.

Selection procedure of Students:

At the beginning of every academic year, the weak students or slow learners are identified through oral discussion and class performance. These students are considered for the enrollment of remedial class.

Syllabus Frame:

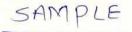
As per the need of students, the syllabus for the remedial classes has been framed and the same is distributed into two parts for B.Sc. I and B.Sc. II year students. 24 hours have been allotted to complete the concerned syllabus of each class (B.sc. I & II Year). The said course syllabus for each class is expected to complete within two months therefore for each class 03 hours per week have been allotted.

Student's progress:

After the completion of syllabus of each class successfully, students progress is evaluated by conducting a test of 20~k=marks comprising 20~multiple choice questions based on the syllabus framed for the remedial class. The result of the students (mark list of tests) is displayed on the notice board after five days from the date of test conducted.

Besides, the faculty of chemistry is always motivating the students to participate in various activities of the college and continuously encourage and cooperate them by counseling.





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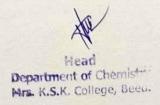
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Department of Chemistry

Analysis report of Student's Feedback for Remedial Classes

For the academic year – 2021-22

Sr. No	Particulars	% of Very Good	% of Good	% of Satisfactory	% of Unsatisfactory
1.	Remedial class course content			-	11-41
2.	Relevance of remedial course content with main courses				
3.	Learning resources for the remedial class (Library, ICT etc.)				
4.	Helpful in improving the subject knowledge				
5.	Various skills inculcated through course				
6.	Assignments/Evaluation transparency				
7.	Interest generated by the teacher				
8.	Extent of coverage of course during prescribed period				



SAMPLE

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Mrs. Kesharbai Sonajirao Kshirsagar alias Kaku Arts, Science and Commerce College, Beed.

Student's Feedback for Remedial Class

The State of the S	Class:	
Department:	Year of Study:	

Make a tick mark in the appropriate cell:

Sr. No	Particulars	Very Good	Good	Satisfactory	Very Good Good Satisfactory Unsatisfactory
·	Remedial class course content				
2.	Relevance of remedial course content with main courses				
3.	Learning resources for the remedial class (Library, ICT etc.)				
4	Ease of the content for conceptual understanding				
5.	Various skills inculcated through course				
.9	Assignments/Evaluation transparency				
7.	Interest generated by the teacher				
00	Extent of coverage of course during prescribed period				

Suggestions for further improvement:

Signature

Head

Department of Chemistry

Mrs. K.S.K. College, Beed,

Head pepartment of Chemistry Mrs. K.S.K. College, Beed

Remedial Class Test Mark Sheet -2019-20

Department of Chemistry

Class: B.Sc. I Year

Sr. No.	Name of Students	Mark Obtained
1	Ade Raving Babarao	09
2	Ahire Anamika Bhagavat	08
3	Bhosicar y- s	10
4	Chavan Ganesh sadam	07
5	phere sancheti sendipan	04
6	Gade Pani lalasabeb	07
7	anikwad pavi Balasaheh	10
8	Hange saleshi sentosh	07
9	Inkar sonket madhucar	09
10	Jagter Amol Hamidas	10
11	Kadam Dadasaheb Anna	08
12	kale Y. K	08
13	mandre prajakta m	08
14	Netke suchin sarjerao	98
15	orthal ristal kailas	10
16	patil Gonesh kailed	06
17	pathod sonal Shesherao	09
18	Sancep Naibhan. Bhagchond	08
19	syed schil quisar	08
20	-pule vijay Ankush	07
21	ward proyonky uddhow	08
22	wagh arpity Abhiman	07
23	ware Harshada Dilip	-20
24	wagh synil Kishan	10
25	Hele. Ashwini Bapy	09
26		

Name of Teacher: Or Khakre P'R

Signature ad Department of Chemistry Mrs. K.S.K. College, Beed.

Remedial Class Test Mark Sheet -2019-20

Department of Chemistry

Class: B.Sc. II Year

Sr. No.	Name of Students	Mark Obtained
1	Agwan Visay Bubasabeb	08
2	Bade phygneshwar Laxman	08
3	Bankar Amol Ashok	10
4	Charan Abhijit Mahader	09
5	chavan posatik Deepak	08
6	cholse Gajanan Ashok	67
7	Dake Rohini Namdeo	89
8	Dolas Kiran Nanybhau	08
9	Gairward Shital Balgsubeb	09
10	Gholap Ganesh Bhaskar	09
11	Ingole Varshy Santosh	09
12	Jagtap Tushar Anil	09
13	Kale Amol Ichandy	10
14	Kale swati Balgsaheb	09
15	Mane Ashwini Gokul.	08
16	Munde Abhishel Dilip	09
17	Mismal Aditi Kishor	09
18	parse mahader Babasaheb	68
19	potdar provin vitthal	09
20	Rauf Marayan sundar	09
21	Sanap shouti madhukar	08
22	Thite shahaji Vaisinath	08
23	Thosar Akshay Mahadev	10
24	Ugale posashant shivmats	08
25	IManve Krushna Ramu	08
26	Zodge proetik Kishanrao	08

Name of Teacher: Dro. S. B. Maulage.

Signature:

Department of Chemistry Remedial Progress Test 2021-22

Head Department of Chemistre Mrs. K.S.K. College, Been

					Inad -
Date:				Marks obta	= 20
				Out of	= 20
Multiple choice questi	ons with one correct	answer.			
1. Isotones have same n	umber of				
a) Protons	b) Electrons	c) Neutrons	d) Protons an	d Neutrons	
The percentage of io	nic character in HCI mol	ecule is approximate	elv		
a) /1%	b) 100% c) 12%	d) 17%	7.740		
The density of nucleu	is is of the order of	105 6 050003			
a) 10 ¹⁴ g/cm ³	b) 1012 g/cm3	c) 1010 g/cm3	d) 108 g/cm3		
 Isotopes have same n 	umber of	satistica astanta			
a) Protons	b) Electrons	c) Neutrons	d) None of the	ese	
Isobars have same					
a) Atomic number	b) Atomic mass	c) Both A and B	d) None of the	ese	
The value of Planck's	s constant h is numericall	ly equal to			
a) ergs sec	b) J sec c) Both A	and B d) Ne	ther A or B		
	of an element is limited	to			
a) 8 b) 9	c) 6 d) 5				
8. The energy of an elec		Steffen Zaer	20		
a) Negative	b) Zero c) Positive	e d) None of the	above		
Covalent compounds			22.2		
a) Polar solvents	b) Non-polar solvents		ne of the above		
	tion (s) for the formation				
	nisation energy of an aton				
	ectron affinity of an aton	n for anion formation	1		
c) Higher lattice ener	gy of the resultant ionic				
c) Higher lattice ener d) All of the above	gy of the resultant ionic				
c) Higher lattice energyd) All of the above11. Which is true about it	gy of the resultant ionic on onic bonds	compound	d kallia a ala	1/21	. 11:
c) Higher lattice ener d) All of the above 11. Which is true about is a) Directional nature	gy of the resultant ionic o onic bonds b) Stereo Isomerism		nd boiling points	s d) Noncrys	talline nature
c) Higher lattice ener d) All of the above 11. Which is true about is a) Directional nature 12. Ionic compounds, in a	gy of the resultant ionic of onic bonds b) Stereo Isomerism general, possess both	c) High melting an	nd boiling points	s d) Noncrys	talline nature
c) Higher lattice ener d) All of the above 11. Which is true about is a) Directional nature 12. Ionic compounds, in a a) High melting point	gy of the resultant ionic of onic bonds b) Stereo Isomerism general, possess both is, boiling points and non	compound c) High melting and the control of the co	nd boiling points	s d) Noncrys	talline nature
c) Higher lattice ener d) All of the above 11. Which is true about is a) Directional nature 12. Ionic compounds, in a a) High melting point b) High melting point	gy of the resultant ionic of onic bonds b) Stereo Isomerism general, possess both its, boiling points and non its and low-boiling points	compound c) High melting and the contractional bonds	nd boiling points	s d) Noncrys	talline nature
c) Higher lattice ener d) All of the above 11. Which is true about is a) Directional nature 12. Ionic compounds, in a a) High melting point b) High melting point c) Directional nature	gy of the resultant ionic of onic bonds b) Stereo Isomerism general, possess both its, boiling points and non its and low-boiling points of bonds and low boiling	compound c) High melting and another control of the control of the compound o	nd boiling points	s d) Noncrys	talline nature
c) Higher lattice ener d) All of the above 11. Which is true about is a) Directional nature 12. Ionic compounds, in g a) High melting point b) High melting point c) Directional nature d) High solubilities in	gy of the resultant ionic of onic bonds b) Stereo Isomerism general, possess both its, boiling points and non its and low-boiling points of bonds and low boiling in polar and non-polar sol	compound c) High melting and the second sec		s d) Noncrys	talline nature
c) Higher lattice ener d) All of the above 11. Which is true about is a) Directional nature 12. Ionic compounds, in a a) High melting point b) High melting point c) Directional nature d) High solubilities in 13. Molten sodium chlori	gy of the resultant ionic of onic bonds b) Stereo Isomerism general, possess both its, boiling points and non its and low-boiling points of bonds and low boiling in polar and non-polar sol- ide conducts electricity d	compound c) High melting and addrectional bonds g points vents ue to the presence of			
c) Higher lattice ener d) All of the above 11. Which is true about is a) Directional nature 12. Ionic compounds, in a) High melting point b) High melting point c) Directional nature d) High solubilities in 13. Molten sodium chlori a) Free electrons	gy of the resultant ionic of conic bonds b) Stereo Isomerism general, possess both its, boiling points and non its and low-boiling points of bonds and low boiling in polar and non-polar solide conducts electricity did b) Free ions	compound c) High melting and addrectional bonds goints vents ue to the presence of c) Free molecules	d) Atoms of s	sodium and ch	
c) Higher lattice ener d) All of the above 11. Which is true about is a) Directional nature 12. Ionic compounds, in a) High melting point b) High melting point c) Directional nature d) High solubilities in 13. Molten sodium chlori a) Free electrons 14. Most predominantly in	gy of the resultant ionic of onic bonds b) Stereo Isomerism general, possess both its, boiling points and non its and low-boiling points of bonds and low boiling polar and non-polar solide conducts electricity db) Free ions ionic compound will be f	compound c) High melting and directional bonds g points vents ue to the presence of c) Free molecules formed by the combiner	d) Atoms of s	sodium and ch	
c) Higher lattice ener d) All of the above 11. Which is true about is a) Directional nature 12. Ionic compounds, in a a) High melting point b) High melting point c) Directional nature d) High solubilities in 13. Molten sodium chlori a) Free electrons 14. Most predominantly is a) 1 and 17	gy of the resultant ionic of conic bonds b) Stereo Isomerism general, possess both ts, boiling points and non ts and low-boiling points of bonds and low boiling polar and non-polar solide conducts electricity db) Free ions ionic compound will be fb) 2 and 16	compound c) High melting and addrectional bonds g points vents ue to the presence of c) Free molecules formed by the combine	d) Atoms of s nation of the gro d) 1 and 18	sodium and ch	
c) Higher lattice ener d) All of the above 11. Which is true about is a) Directional nature 12. Ionic compounds, in a a) High melting point b) High melting point c) Directional nature d) High solubilities in 13. Molten sodium chlori a) Free electrons 14. Most predominantly is a) 1 and 17 15. Of the following solv	gy of the resultant ionic of conic bonds b) Stereo Isomerism general, possess both ts, boiling points and non ts and Iow-boiling points of bonds and Iow boiling polar and non-polar solide conducts electricity db) Free ions ionic compound will be fb) 2 and 16 ents, the one most likely	compound c) High melting and adirectional bonds goints vents ue to the presence of c) Free molecules formed by the combine c) 3 and 15 to dissolve ionic con	d) Atoms of s nation of the gro d) 1 and 18 npounds is	sodium and ch	
c) Higher lattice ener d) All of the above 11. Which is true about is a) Directional nature 12. Ionic compounds, in a a) High melting point b) High melting point c) Directional nature d) High solubilities in 13. Molten sodium chlori a) Free electrons 14. Most predominantly is a) 1 and 17 15. Of the following solv a) Carbon tetrachloric	gy of the resultant ionic of onic bonds b) Stereo Isomerism general, possess both its, boiling points and non its and low-boiling points of bonds and low boiling in polar and non-polar solide conducts electricity db) Free ions ionic compound will be fb) 2 and 16 ents, the one most likely its by Methar	compound c) High melting and addrectional bonds g points vents use to the presence of c) Free molecules formed by the combine c) 3 and 15 to dissolve ionic comol	d) Atoms of s nation of the gro d) 1 and 18 npounds is d) Benzene	sodium and ch	
c) Higher lattice ener d) All of the above 11. Which is true about is a) Directional nature 12. lonic compounds, in a a) High melting point b) High melting point c) Directional nature d) High solubilities in 13. Molten sodium chlori a) Free electrons 14. Most predominantly is a) 1 and 17 15. Of the following solv a) Carbon tetrachloric 16. The pair of elements	gy of the resultant ionic of conic bonds b) Stereo Isomerism general, possess both ts, boiling points and non ts and low-boiling points of bonds and low boiling polar and non-polar solide conducts electricity db) Free ions ionic compound will be fb) 2 and 16 ents, the one most likely deb) Methar which form a compound	compound c) High melting and adirectional bonds g points vents ue to the presence of c) Free molecules formed by the combine c) 3 and 15 to dissolve ionic comol c) Water with maximum ionic	d) Atoms of s nation of the gro d) 1 and 18 npounds is d) Benzene e character is	sodium and ch	
c) Higher lattice ener d) All of the above 11. Which is true about is a) Directional nature 12. Ionic compounds, in a a) High melting point b) High melting point c) Directional nature d) High solubilities in 13. Molten sodium chloria a) Free electrons 14. Most predominantly is a) 1 and 17 15. Of the following solva) Carbon tetrachloric 16. The pair of elements a) Na and O	gy of the resultant ionic of onic bonds b) Stereo Isomerism general, possess both ts, boiling points and non ts and low-boiling points of bonds and low boiling polar and non-polar solide conducts electricity db) Free ions ionic compound will be fb) 2 and 16 ents, the one most likely deb) Methar which form a compound b) Na and I	compound c) High melting and addrectional bonds g points vents use to the presence of c) Free molecules formed by the combine c) 3 and 15 to dissolve ionic comol	d) Atoms of s nation of the gro d) 1 and 18 npounds is d) Benzene	sodium and ch	
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Date: 25/10/2021

To,

The Principal, Mrs. K. S. K. Arts, Sci. & Comm. College, Beed.

Sub: Permission for remedial coaching classes - Chemistry department.

Respected Madam,

Faculty of chemistry department has identified the weak students / slow learners in subject through oral discussion and classroom performance. These students are needed to be considered for the enrollment of remedial classes. Department of chemistry, therefore is willing to conduct remedial coaching during 01/11/2021 to 11/01/2022 for identified slow learners from B.Sc. I & II year classes so as to bring them in the mainstream of average learners in the chemistry subject.

Therefore you are requested to give permission to start the remedial coaching by the department of Chemistry.

Thanking you.

Yours Faithfully

Dr. P.R. Khakre Head, Dept. of Chemistry

Department of Chemistry Mrs. K.S.K. College, Beed.

Best Practice -4

1. Title of the Practice: 'Counseling and Guidance Committees for Admission Process'

2. Objective of the Practice:

To adopt more convenient and transparent admission process by constituting 'Counseling and Guidance Committees for Admission Process'.

3. The Context:

Admission process of any higher education institution is somewhat a critical task specially in case of online admission process. Nevertheless, if it is well planned and executed professionally then satisfaction level of prospective students increases and it also helps in branding of the institute. Here is an attempt to improve the admission process of the institute so as to make it more accessible for socially and economically deprived sections of society.

4. The Practice:

The college offers various certificate courses, diploma courses, undergraduate courses and postgraduate courses. The college displays an advertisement of admission process for each course in the local and regional newspapers and in the campus area well in advance.

Currently a few students are against online admission process due to lack of detailed knowledge of the system and unaware of how to use it effectively. So when students approach to college, then they are helpless in most of the situations and it creates negative attitude for technology usage to simplify such critical process. To overcome such issues, a 'Counseling and Guidance Committees for Admission Process' is constituted for the better convenience of the students in assisting the online application process and also choosing the right courses and selecting desired subjects for various courses. For each given course, a separate committee is constituted and members of each committee counsel and guide almost every student who is aspired to take admission in our college. Aspired students first apply online on the college website to register their names for admission. For the online registration, College makes available a well equipped computer lab for students where staff assists the students in applying the online registration. After registration, students collect prospectus from the administrative office of the college. College prospectus provides all the required information concerning to the various certificate courses, diploma courses, undergraduate courses and post graduate courses as well as rules and regulations of the

college. The college adopts 'First come first serve' policy by maintaining the reservation policy of the government for the admission process. The college has no management quota to maintain the transparency in admission process. Invalid applications are sorted out by the screening committee and the list of valid application or selected students for different courses is displayed on the college notice board. After being selected students are again assisted by the committee members to apply online on the university website.

5. Evidence of Success;

College assists students from their registration to till his/her admission is completed and maintains complete transparency in admission process as a result of which college is highly appreciated by students, parents and other stakeholders of the college at large. Because of this practice, we realize that, the percentage of changing subjects by the students after admission is minimized upto the great margin and students are also benefited with the counseling to choose the correct subjects as per their like and interest. Most importantly, students in the nearby regions preferring this college at the top for their admission. Moreover, this practice has also helped the students in getting detailed information of the college like sports, library, language laboratory, computer laboratory, activities to be run and other facilities in the college campus well in advance before their admission and eventually College marketing is also done through this practice.

6. Problems encountered:

The members of the admission committee face various problems while working on the admission process. Some of the major problems faced by the institution while implementing the online problems are as under.

Computer illiteracy among the students makes them indifferent to approach online admission process. It takes much time to counsel and guide every student about the process.

Rural area students find the procedure difficult and inconvenient ever after the guidance by the committee. The students request the committee members to fill up the online form on their behalf.

Many times slow, internet speed and data problems create hindrance in the admission process.

Resources required:

To run this practice a computer laboratory and faculty availability are required. Cooperation and support of college staff is necessary for the transparent admission process to operate.

नवगण शिक्षण संस्था राजुरी (न.) चे

सौ.केशरबाई सोनाजीराव क्षीरसागर उर्फ काकू कला, वाणिज्य व विज्ञान महाविद्यालय, बीड शैक्षणिक वर्ष २०२०-२०२१ ऑनलाईन प्रवेश सुचना ONLINE ADMISSION NOTICE

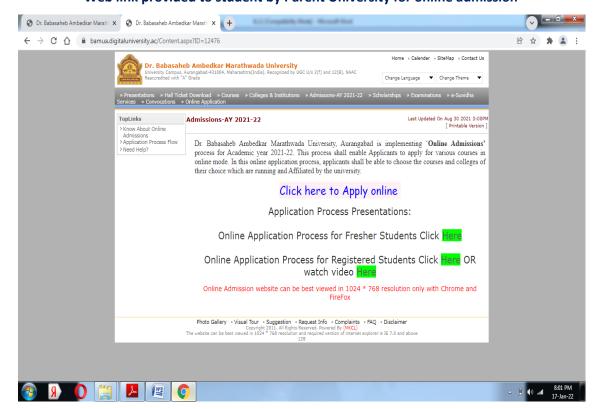
अकरावी, बारावी (कनिष्ट विभाग) बी. ए., बी. एस्सी व बी. कॉम. (पदवी विभाग) तसेच एम.ए., एम. एस्सी व एम. कॉम. (पदव्युत्तर विभाग) अभ्यासक्रमाच्या (Online Admission) ऑनलाईन प्रवेशप्रक्रियेसाठी महाविद्यालयाच्या www.kskcollegebeed.com या वेबसाईटवर जावे. त्यासाठीचे पुढील मार्गदर्शक सुचना (टप्पे) खालील प्रमाणे आहेत.

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११. अकरा प्रवेशा	किनष्ठ महाविद्द शुल्क योजनेचे ऑनलाईन भरण भरावेत. वी व बारावी साठी	ालयातील EBC ः अर्ज ऑनलाईन भरु यात येऊन त्याची एव ११ बी व १२ बी विज्ञा प्रा.निसंगंध एम.डी.	न कार्यालयात दाखल कर क प्रत कार्यालयात सादर न ११ बी व १२ बी क प्रा.नेटके डी.ई. मो. ९	ावेत व भारत स करावी. <u>WW</u> ला,वाणिज्य १४२१३४८४८३	एम.सी.स	शिष्यवृत्ती (GOI) व ahadbt.con ही.सी. व्यवसाय अभ्यासक	व इतर शिष्यवृत्तीच सर्व अ 1 या वेबसाईटवर अर्ज
थेर. अकरा अकरा अवेज्ञा-	किनष्ठ महाविद्द शुल्क योजनेचे ऑनलाईन भरण भरावेत. वी व वासवी साठी	ग्रालयातील EBC र अर्ज ऑनलाईन भरू यात येऊन त्याची एव ११ वी व १२ वी विज्ञा प्रा.निसंगंध एम.डी. को.९९२१०७५३१५ प्रा.मांडवे पी.बी.	न कार्यालयात दाखल कर क प्रत कार्यालयात सादर न ११ बी व १२ बी क प्रा.नेटके डी.ई. मो. ९	ावेत व भारत स करावी. <u>WW</u> ला,वाणिज्य १४२१३४८४८३	एम.सी.व् प्रा.वंजा	शिष्यवृत्ती (GOI) व ahadbt.con ही.सी. व्यवसाय अभ्यासक	व इतर शिष्यवृत्तीच सर्व अ 1 या वेबसाईटवर अर्ज जम हप्हप्हुप्
थकरा अकरा अकरा अवेशाः बी.ए.	किनष्ठ महाविद्द शुल्क योजनेचे ऑनलाईन भरण भरावेत. वी व वासवी साठी	ग्रालयातील EBC र अर्ज ऑनलाईन भरु यात येऊन त्याची एर ११ वी व १२ वी विका प्रा.निसंगंध एम.डी. मो.९९२१०७५३१५ प्रा.मांडवे पी.बी. मो.९७६३३६४५४३ प्रा.पवार एस.एस. प्रा.पोटे वी.टी.	न कार्यालयात दाखल कर क प्रत कार्यालयात सादर न ११ वी व १२ वी क प्रा.नेटके डी.ई. मो. ९ प्रा.घुमरे ए.ए. मो.९	ावेत व भारत स करावी. <u>WW</u> ला,वाणिज्य १४२१३४८४८३ ४२००१३१६१	एम.सी. प्रा.वंजा प्रा.वंजा डॉ.देवव प्रा.कटा	शिष्यवृत्ती (GOI) व ahadbt.con ही.सी. व्यवसाय अभ्यासक रे व्ही.एस. मो. ९६६ आणकर एस.एस. मो.९४३ रे बी.व्ही. मो.९४३	व इतर शिष्यवृत्तीच सर्व अ 1 या वेबसाईटवर अर्ज जम हप्हप्हुप्
११. अकरा व्येशाः वी.ए.र्	कनिष्ठ महाविद्द शुल्क योजनेचे ऑनलाईन भरण भरावेत. वी व वासवी साठी	ग्रालयातील EBC व अर्ज ऑनलाईन भरुन यात येऊन त्याची एव ११ वी व १२ वी विज्ञा प्रा.निसंगंध एम.डी. मो.९९२१०७५३१५ प्रा.मांडवे पी.बी. मो.९७६३३६४५४३ प्रा.पवार एस.एस. प्रा.पोटे बी.टी. डॉ.शिंदे एस.एन.	न कार्यालयात दाखल कर क प्रत कार्यालयात सादर न ११ वी व १२ वी क प्रा.नेटके डी.ई. मो. ९ प्रा.घुमरे ए.ए. मो.९ मो.९४०४००५८६५ मो.९४२३४६९५६४	ावेत व भारत स करावी. <u>WW</u> ला,वाणिज्य १४२१३४८४८३ ४२००१३१६१	एम.सी. प्रा.वंजारे डॉ.देवव	शिष्यवृत्ती (GOI) व ahadbt.con ही.सी. व्यवसाय अभ्यासक रे व्ही.एस. मो. ९६६ प्राणकर एस.एस. मो.९४२ रे बी.व्ही. मो.९९२	व इतर शिष्यवृत्तीच सर्व अ व या वेबसाईटवर अर्ज व ६५६५९७५१ २२२९५३१४
११. अकराः वी.ए. वी.ए. वी.ए.	कनिष्ठ महाविद्द शुल्क योजनेचे ऑनलाईन भरण भरावेत. वी व वारावी साठी प्रथम वर्ष हतीव वर्ष तृतीव वर्ष सी.प्रथम वर्ष	ालयातील EBC व अर्ज ऑनलाईन भरुन यात येऊन त्याची एव ११ वी व १२ वी विज्ञा प्रा.निसंगंध एम.डी. मो.९९२१०७५३१५ प्रा.मांडवे पी.बी. मो.९७६३३६४५४३ प्रा.पवार एस.एस. प्रा.पोटे बी.टी. डॉ.शिंदे एस.एन. प्रा.गायकवाड एस.व्ही	न कार्यालयात दाखल कर क प्रत कार्यालयात सादर न ११ वी व १२ वी क प्रा.नेटके डी.ई. मो. ९ प्रा.घुमरे ए.ए. मो.९ मो.९४०४००५८६५ मो.९४२३४६९५६४	ावेत व भारत स करावी. <u>WW</u> ला,वाणिज्य १४२१३४८४८३ ४२००१३१६१	एम.सी. प्रम.सी. प्रा.वंजां डॉ.देवव प्रा.कटां डॉ.शिंदे प्रा.राकर	शिष्यवृत्ती (GOI) व ahadbt.com ही.सी. व्यवसाय अभ्यासक हे व्ही.एस. मो. ९६६ आणकर एस.एस. मो. ९६६ रे बी.व्ही. मो.९९६ ए.व्ही. मो.९९६	व इतर शिष्यवृत्तीच सर्व अ व या वेबसाईटवर अर्ज व ६५६५९७५१ २२२९५३१४ २२७८९११५
११. अकरा वी.ए. वी.ए. वी.ए. वी.एस वी.एस	कनिष्ठ महाविद्द शुल्क योजनेचे ऑनलाईन भरण भरावेत. वी व वारावी साठी प्रथम वर्ष ह्वतीय वर्ष तृतीय वर्ष सी.प्रथम वर्ष सी.प्रथम वर्ष	ग्रालयातील EBC व अर्ज ऑनलाईन भरुन यात येऊन त्याची एव ११ बी व १२ बी विज्ञा प्रा.निसंगंध एम.डी. मो.९९२१०७५३१५ प्रा.मांडवे पी.बी. मो.९७६३३६४५४३ प्रा.पवार एस.एस. प्रा.पोटे बी.टी. डॉ.शिंदे एस.एन. प्रा.गायकवाड एस.व्ही डॉ.शेळके ए.एन.	न कार्यालयात दाखल कर क प्रत कार्यालयात सादर न ११ वी व १२ वी क प्रा.नेटके डी.ई. मो. ९ प्रा.घुमरे ए.ए. मो.९ मो.९४०४००५८६५ मो.९४२३४६९५६४	ावेत व भारत स करावी. <u>WW</u> ला,वाणिज्य १४२१३४८४८३ ४२००१३१६१	एम.सी. प्रा.वंजा डॉ.देवव प्रा.कटा डॉ.प्रिंदे प्रा.राऊट	शिष्यवृत्ती (GOI) व ahadbt.com ही.सी. व्यवसाय अभ्यासक र व्ही.एस. मो. ९६१ अणकर एस.एस. मो. ९६१ र वी.व्ही. मो. ९९१ ए.व्ही. मो. ९०१ त एस.आर. मो. ९४	व इतर शिष्यवृत्तीच सर्व अ या वेबसाईटवर अर्ज स्व है । स्व है । स्व है । स्व है । स्व है । स्व है । स्व है । स्व ह
११. अकरा प्रवेजाः वी.ए. वी.एस वी.एस वी.एस वी.एस	कनिष्ठ महाविद्द शुल्क योजनेचे ऑनलाईन भरण भरावेत. वी व वारावी साठी प्रथम वर्ष ह्वतीय वर्ष स्त्री.प्रथम वर्ष सी.प्रथम वर्ष सी. द्वितीय वर्ष सी. द्वितीय वर्ष	ालयातील EBC व अर्ज ऑनलाईन भरुन यात येऊन त्याची एव ११ बी व १२ बी विज्ञा प्रा.निसंगंध एम.डी. मो.९९२१०७५३१५ प्रा.मांडये पी.बी. मो.९७६३३६४५४३ प्रा.पवार एस.एस. प्रा.पोटे बी.टी. डॉ.शिंदे एस.एन. प्रा.गायकवाड एस.व्ही डॉ.शेळके ए.एन. डॉ.खाकरे पी.आर.	न कार्यालयात दाखल कर क प्रत कार्यालयात सादर न ११ वी व १२ वी क प्रा.नेटके डी.ई. मो.९ प्रा.घुमरे ए.ए. मो.९ मो.९४०४००५८६५ मो.९६५७४२५३७४ मो.९४२३४६९५६४	ावेत व भारत स करावी. <u>WW</u> ला,वाणिज्य १४२१३४८४८३ ४२००१३१६१	एम.सी. प्रा.वंजा डॉ.देवव प्रा.कटा डॉ.प्रिंदे प्रा.राऊट	शिष्यवृत्ती (GOI) व ahadbt.com ही.सी. व्यवसाय अभ्यासक हे व्ही.एस. मो. ९६६ आणकर एस.एस. मो. ९६६ रे बी.व्ही. मो.९९६ ए.व्ही. मो.९९६	व इतर शिष्यवृत्तीच सर्व अ या वेबसाईटवर अर्ज स्व है । स्व है । स्व है । स्व है । स्व है । स्व है । स्व है । स्व ह
थ्रेर अकरा प्रवेशाः बी.ए. बी.एस बी.एस बी.एस बी.एस बी.एस	कनिष्ठ महाविद्द शुल्क योजनेचे ऑनलाईन भरण भरावेत. वी व वारावी साठी प्रथम वर्ष ह्वतीय वर्ष तृतीय वर्ष सी.प्रथम वर्ष सी.प्रथम वर्ष	ग्रालयातील EBC व अर्ज ऑनलाईन भरुन यात येऊन त्याची एव ११ बी व १२ बी विज्ञा प्रा.निसंगंध एम.डी. मो.९९२१०७५३१५ प्रा.मांडवे पी.बी. मो.९७६३३६४५४३ प्रा.पवार एस.एस. प्रा.पोटे बी.टी. डॉ.शिंदे एस.एन. प्रा.गायकवाड एस.व्ही डॉ.शेळके ए.एन.	न कार्यालयात दाखल कर क प्रत कार्यालयात सादर न ११ वी व १२ वी क प्रा.नेटके डी.ई. मो. ९ प्रा.घुमरे ए.ए. मो.९ मो.९४०४००५८६५ मो.९६५७४२५३७४ मो.९४२३४६९५६४ मो.९४२१३४२८०	ावेत व भारत स करावी. <u>WW</u> ला,वाणिज्य १४२१३४८४८३ ४२००१३१६१	एम.सी. प्रा.वंजा डॉ.देवव प्रा.कटा डॉ.प्रिंदे प्रा.राऊट	शिष्यवृत्ती (GOI) व ahadbt.com ही.सी. व्यवसाय अभ्यासक रे व्ही.एस. मो.९६१ रे वी.व्ही. मो.९९१ ए.व्ही. मो.९०१ त एस.आर. मो.९४ लगे एस.बी. मो.९४	व इतर शिष्यवृत्तीच सर्व अ या वेबसाईटवर अर्ज स्व है स्व
११. अकरा प्रवेशाः बी.ए. बी.एस बी.एस बी.एस बी.एस बी.एस एम.ए.	किनष्ठ महाविद्द शुल्क योजनेचे ऑनलाईन भरण भरावेत. वी व वारावी साठी प्रथम वर्ष ह्वतीय वर्ष सी. प्रथम वर्ष सी. तृतीय वर्ष म. प्रथम, व तृतीय वर्ष	ालयातील EBC व अर्ज ऑनलाईन भरुन यात येऊन त्याची एव ११ बी व १२ बी विज्ञा प्रा.निसंगंध एम.डी. मो.९९२१०७५३१५ प्रा.मांडये पी.बी. मो.९७६३३६४५४३ प्रा.पवार एस.एस. प्रा.पोटे बी.टी. डॉ.शिंदे एस.एन. प्रा.गायकवाड एस.व्ही डॉ.शेळके ए.एन. डॉ.खाकरे पी.आर.	न कार्यालयात दाखल कर क प्रत कार्यालयात सादर न ११ वी व १२ वी क प्रा.नेटके डी.ई. मो. ९ प्रा.घुमरे ए.ए. मो.९ मो.९४०४००५८६५ मो.९४५३४६९५६४ मो.९४०८५५७४१ मो.९४०५६७१००६	ावेत व भारत स करावी. <u>WW</u> ला,वाणिज्य १४२१३४८४८३ ४२००१३१६१	एम.सी.१ प्रा.कंजां डॉ.देवव प्रा.कंजां डॉ.हंशदे प्रा.राज डॉ.माऊ डॉ.माऊ डॉ.माऊ	शिष्यवृत्ती (GOI) व ahadbt.com ही.सी. व्यवसाय अभ्यासक रे व्ही.एस. मो.९६६ ठाणकर एस.एस. मो.९६६ रे बी.व्ही. मो.९९६ ए.व्ही. मो.९०। त एस.आर. मो.९४ लगे एस.बी. मो.९४ एन.पी. मो.९४	व इतर शिष्यवृत्तीच सर्व अ व या वेबसाईटवर अर्ज स्व है स्व है स्
११. अकरा बी.ए. बी.एस बी.एस बी.एस बी.एस बी.एस प्रम.ए. प्रथम, (१	किनष्ठ महाविद्द शुल्क योजनेचे ऑनलाईन भरण भरावेत. बी व बारावी साठी प्रथम वर्ष हृतीय वर्ष सी. हृतीय वर्ष सी. हृतीय वर्ष म. प्रथम, व तृतीय वर्ष	ालयातील EBC व अर्ज ऑनलाईन भरुन यात येउन त्याची एव ११ बी व १२ बी विज्ञा प्रा.निसंगंध एम.डी. मो.९५२१०७५३१५ प्रा.मांडवे पी.बी. मो.९७६३३६४५४३ प्रा.पवार एस.एस. प्रा.पोटे बी.टी. डॉ.शिंदे एस.एन. प्रा.गायकवाड एस.व्हीं डॉ.शेळके ए.एन. डॉ.खाकरे पी.आर. डॉ.जाधव एस.एस.	न कार्यालयात दाखल कर क प्रत कार्यालयात सादर न ११ वी व १२ वी क प्रा.नेटके डी.ई. मो. १ प्रा.घुमरे ए.ए. मो.९ मो.९४०४००५८६५ मो.९४५७४२५३७४ मो.९४२३४६९५६४१ मो.९४२१३४२८०० मो.९४२१७५९६४९	ावेत व भारत स करावी. <u>WW</u> ला,वाणिज्य १४२१३४८४८३ ४२००१३१६१	एम.सी.१ प्रा.वंजां डॉ.देवव प्रा.कटां डॉ.शिंदे प्रा.राक्त डॉ.माक डॉ.महेश डॉ.महेश डॉ.नुझह	शिष्यवृत्ती (GOI) व ahadbt.con ही.सी. व्यवसाय अभ्यासक रे व्ही.एस. मो.९६६ ठाणकर एस.एस. मो.९६६ रे बी.व्ही. मो.९९६ ए.व्ही. मो.९९६ ए.व्ही. मो.९९६ त एस.आर. मो.९४ लगे एस.बी. मो.९४ एम.पॉ. मो.९४	व इतर शिष्यवृत्तीच सर्व अ व या वेबसाईटवर अर्ज व स्पर्धप्रध्य २२२९५३१४ २२५८९११५ ७५८०३००० २२६३०९१० २२६३०९१० १२६३०९१० १२६३०९१०
थेर. अकरा बी.ए. बी.ए. बी.ए. बी.एस बी.एस बी.एस प्रथम, प्रथम, प्रथम, प्रथम, प्रथम,	किनष्ठ महाविद्द शुल्क योजनेचे ऑनलाईन भरण भरावेत. बी व बारावी साठी प्रथम वर्ष हितीय वर्ष सी. हितीय वर्ष सी. जुतीय वर्ष म. प्रथम, व तृतीय वर्ष स्ती. हितीय वर्ष स्ती.	ालयातील EBC अर्ज ऑनलाईन भरु- यात येउन त्याची एव ११ वी व १२ वी विज्ञा प्रा.निसंगंध एम.डी. मो.९५२१०७५३१५ प्रा.मांडवे पी.बी. मो.९७६३३६४५४३ प्रा.पांटे वी.टी. डॉ.शिंदे एस.एन. प्रा.गायकवाड एस.व्हीं डॉ.शेळके ए.एन. डॉ.खाकरे पी.आर. डॉ.जाधव एस.एस.	न कार्यालयात दाखल कर क प्रत कार्यालयात सादर व ११ वी व १२ वी क प्रा.नेटके डी.ई. मो.९ प्रा.घुमरे ए.ए. मो.९ मो.९४०४००५८६५ मो.९४५७४२५३७४ मो.९४५३४२८०० मो.९४०५६७१००६ मो.९४२१३५६२८० मो.९४२१३३६२२८	वित व भारत स करावी. WW ला,वाणिज्य ४४२१३४८४८३ ४२००१३१६१	एम.सी.१ प्रा.वंजां डॉ.देवव प्रा.कटां डॉ.शिंदे प्रा.राक्त डॉ.माक डॉ.महेश डॉ.महेश डॉ.नुझह	शिष्यवृत्ती (GOI) व ahadbt.con ही.सी. व्यवसाय अभ्यासक रे व्ही.एस. मो.९६६ ठाणकर एस.एस. मो.९६६ रे बी.व्ही. मो.९९६ ए.व्ही. मो.९९६ ए.व्ही. मो.९९६ त एस.आर. मो.९४ एम.ची. मो.९४ एम.पॉ. मो.९४ एस.एस. से.९१	व इतर शिष्यवृत्तीच सर्व अ व या वेबसाईटवर अर्ज व व व व व व व व व व व व व व व व व व व

Web link provided to student on College website for online admission



Web link provided to student by Parent University for online admission



Practice-5. The college publishes the college magazine – '*DURWANKUR*' every year so as to provide a platform for college students to nourish and develop their hidden talent of writing articles on various issues under the guidance of teachers.

Practice-6.College Conducts 'Science exhibition' every year for the students to imbibe scientific temperament among students and also to create liking for science.

Practice-7.College organizes 'One Night for Senior Citizens' program every year for all the senior citizens of the Beed city to perform on the stage.

Practice-8. 'School-College students' interactions' are conducted every year. The college invites students of nearby schools and offers them interaction with our college students. College students perform some science experiments and demonstrate various concepts of science.

Practice-9. Department of Chemistry of our college conducts **National Chemistry Quiz Competitions** wherein students from different states of the India participate by a number more than a thousand.

All the information mentioned above in this file is true, hence certified.

Co-Ordinator Internal Quality Assurance Cell Idrs. K.S.K. College, Beed, (M.S.) N.S.S.R.(N.) Mrs.Kesharbai Sonajirao Kshirsagar Alias Kaku Arts, Science and Commerce College, Beed.